Lecture-07: The Thermodynamic Limit

1 The thermodynamic limit

The main purpose of statistical physics is to understand the macroscopic behaviour of a large number, $N \gg 1$, of microscopic components (atoms, molecules, etc.) under simple local interactions. For example, in the case of water in a bottle, the number of H_2O molecules N is typically of order 10^{23} (18g of water contains approximately 6×10^{23} molecules), and this huge number leads physicists to focus on the $N \to \infty$ limit, also called **the thermodynamic limit**.

1.1 The intensive thermodynamic potentials

For large N, the thermodynamic potentials are proportional to N. The **intensive thermodynamic potentials** $f(\beta), u(\beta), s(\beta)$ are defined as follows.

Definition 1.1 (Intensive thermodynamic potentials). Denoting the thermodynamic potentials for N particle system as $F_N(\beta), U_N(\beta), S_N(\beta)$ for the free energy, the internal energy, and the canonical entropy respectively. We can define the **free energy density**, the **energy density**, and the **entropy density** as

$$f(\beta) = \lim_{N \to \infty} \frac{F_N(\beta)}{N}, \qquad u(\beta) = \lim_{N \to \infty} \frac{U_N(\beta)}{N}, \qquad s(\beta) = \lim_{N \to \infty} \frac{S_N(\beta)}{N}. \qquad (1)$$

Partition function $Z(\beta)$ is a sum of exponentials, and hence is smooth and analytic. It follows that the free energy $F_N(\beta) = -\frac{1}{\beta} \ln Z(\beta)$ is also analytic.

Definition 1.2 (Phase transition). We say that a **phase transition** occurs, whenever the free energy density $f(\beta)$ is non-analytic.

Since the free entropy density $\phi(\beta) = -\beta f(\beta)$ is convex, the free energy density is necessarily continuous whenever it exists. The phase transitions correspond to qualitative changes in the underlying physical system.

Definition 1.3 (Types of singularities). Often, the non-analyticities occur at isolated points say β_c.

- The free energy density is continuous, but its derivative with respect to β is discontinuous at β_c. This singularity is called a **first-order phase transition**.

- The free energy and its first derivative are continuous, but the second derivative is discontinuous at β_c. This is called a **second-order phase transition**.

1.2 Energy spectrum and Micro-canonincal entropy density

When the number of particles N grows, the volume of the configuration space increases exponentially, i.e. $|\mathcal{X}^N| = |\mathcal{X}|^N$. We have seen before that the system is likely to be found in lowest-energy configurations with high probability at low temperatures. From the definition of Boltzmann distribution, it is easy to check that conditioned on the system to be at certain energy level, it is equally likely to be in any configuration with equal energy. Therefore, one of the important factor of interest is the number of configurations for any given energy level. This information is given by the **energy spectrum** of the system, defined by the set of states with energy in the interval $[E, E + \Delta)$,

$$\Omega_\Delta(E) \triangleq \{x \in \mathcal{X}^N : E \leqslant E(x) < E + \Delta\}.$$

The number of states in $\Omega_\Delta(E)$ is given by $\mathcal{N}_\Delta(E) = |\Omega_\Delta(E)|$.

Definition 1.4 (Equality in leading exponential order). We say that two quantities A_N and B_N are equal **to leading exponential order**, if

$$\lim_{N\to\infty} \frac{1}{N} \log \frac{A_N}{B_N} = 0.$$

We denote this equality by $A_N \doteq_N B_N$.

The energy spectrum diverges exponentially in many systems as $N \to \infty$, if the energy is scaled linearly with N.

Definition 1.5 (Micro-canonical entropy density). More precisely, there exists a function $s(e)$ called the **micro-canonical entropy density**, such that given two numbers e and $\delta > 0$,

$$s(e) = \lim_{N\to\infty} \frac{1}{N} \log \mathcal{N}_{N\delta}(Ne) = \sup_{e' \in [e, e+\delta]} s(e'). \tag{2}$$

Using this notation, we can write the following equality for micro-canonical entropy density

$$\mathcal{N}_\Delta(E) \doteq_N e^{Ns(E/N)}. \tag{3}$$

The micro-canonical entropy density $s(e)$ conveys a great amount of information about the system, and is directly related to the intensive thermodynamic potentials through a fundamental relation.

Remark 1.6. Recall that energy function $E : \mathcal{X}^N \to \mathbb{R}$, and hence we can divide the energy levels into $N\Delta$ intervals. Then, we can partition the configuration space into configurations with energy level in one of these durations. Specifically, we can define

$$\Omega_{N\Delta}(Nk\Delta) = \left\{ x \in \mathcal{X}^N : k\Delta \leqslant \frac{E(x)}{N} < (k+1)\Delta \right\}.$$

Clearly, $(\Omega_{N\Delta}(Nk\Delta) : k \in \mathbb{Z})$ partition the configuration space $\mathcal{X}^N$, and each of these partitions have cardinality $|\Omega_{N\Delta}(Nk\Delta)| = \mathcal{N}_{N\Delta}(Nk\Delta)$. Therefore, we can write the partition function as

$$Z_N(\beta) = \sum_{x \in \mathcal{X}^N} e^{-\beta E(x)} = \sum_{k=-\infty}^{\infty} e^{-\beta Nk\Delta} \mathcal{N}_{N\Delta}(Nk\Delta) \left(\frac{1}{\mathcal{N}_{N\Delta}(Nk\Delta)} \sum_{x \in \Omega_{N\Delta}(Nk\Delta)} e^{-\beta N(\frac{E(x)}{N} - k\Delta)} \right).$$

From equation (3), we get the following equality in leading exponential order $\mathcal{N}_{N\Delta}(Nk\Delta) \doteq_N e^{Ns(k\Delta)}$. If $\mathcal{X}$ is discrete and Δ is small enough so that the energy levels are exactly at $N\Delta$ intervals, then we can write the partition function as

$$Z(\beta) \doteq_N \sum_{k=-\infty}^{\infty} e^{N(s(k\Delta) - \beta k\Delta)}.$$

For continuous energy levels, we can show that $Z_N(\beta) \doteq_N \int e^{N(s(e) - \beta e)} de$, by taking limit of $\Delta \to 0$.

Proposition 1.7. *If the micro-canonical entropy density* (2) *exists for any e and if the limit in equation* (2) *is uniform in e, then the free entropy density* (1) *exists and is given by*

$$\phi(\beta) = \max_e (s(e) - \beta e). \tag{4}$$

If the maximum of $s(e) - \beta e$ is unique, then the internal-energy density equals $\arg\max(s(e) - \beta e)$.

Proof. From the definition, the free entropy density $\phi(\beta)$ can be written as

$$\phi(\beta) = \lim_{N\to\infty} \frac{1}{N} \log(Z_N(\beta)).$$

From the computation of N-particle partition function in Remark 1.6, we can evaluate this limit for discrete configuration space as

$$\phi(\beta) = \lim_{N\to\infty} \frac{1}{N} \sum_{k=-\infty}^{\infty} \exp[N(s(k\Delta) - \beta k\Delta)] = \sup_k [s(k\Delta) - \beta k\Delta] = \max_e [s(e) - \beta e].$$

$\square$

Example 1.8 (N identical two-level systems). We consider an N particle system, where the configuration space of each particle is identical $\mathcal{X} = \{1,2\}$ and consisting of two-levels. That is, $\mathcal{X}^N = \mathcal{X} \times \mathcal{X} \times \cdots \times \mathcal{X}$. For any configuration $x \in \mathcal{X}^N$, we let $x_i \in \mathcal{X}$ denote the configuration of ith particle. As in the previous two-level system example, we let

$$E_{single}(x_i) = \epsilon_1 \mathbb{1}_{\{x_i=1\}} + \epsilon_2 \mathbb{1}_{\{x_i=1\}},$$

where we assume $\epsilon_2 > \epsilon_1$ without any loss of generality and define the energy gap as $\Delta = \epsilon_2 - \epsilon_1$. We take the energy of the N particle system to be the sum of the single-particle energies, i.e.

$$E(x) = E_{single}(x_1) + E_{single}(x_2) + \ldots + E_{single}(x_N).$$

We can next study the energy spectrum for this model. For any configuration $x \in \mathcal{X}^N$, we can define the set of particles in state $k \in \{1,2\}$ as

$$S_k = \{i \in [N] : x_i = k\}.$$

We have $S_1 \cup S_2 = [N]$ and $S_1 \cap S_2 = \varnothing$. If $|S_2| = n$, then the system energy is given by

$$E(x) = \epsilon_1(N - n) + \epsilon_2 n = N\epsilon_1 + n\Delta.$$

The number of possible subsets $S_2 \subseteq [N]$ such that $|S_2| = n$ is equal to the binomial coefficient $\binom{N}{n}$. Therefore, we conclude that $E(x) \in \{E + n\Delta : n = 0, \ldots, N\}$ and for any energy $E = N\epsilon_1 + n\Delta$, there are $\binom{N}{n}$ configurations x such that $E(x) = E$. This is one of the rare examples, where we can completely specify the number of configurations at each energy level, which is

$$\mathcal{N}_\Delta(E) = \binom{N}{n} \approx 2^{N\mathcal{H}(\frac{n}{N})} = 2^{N\mathcal{H}\left(\frac{E-N\epsilon_1}{N\Delta}\right)} = 2^{N\mathcal{H}\left(\frac{\frac{E}{N}-\epsilon_1}{\Delta}\right)}.$$

Using the definition of micro-canonical entropy density (2), we get

$$s(e) = \mathcal{H}\left(\frac{e - \epsilon_1}{\Delta}\right).$$

We can write the free energy density in terms of micro-canonical entropy density as

$$f(\beta) = -\frac{1}{\beta}\phi(\beta) = -\frac{1}{\beta}\sup_e(s(e) - \beta e) = -\frac{1}{\beta}\sup_e\left(\mathcal{H}\left(\frac{e - \epsilon_1}{\Delta}\right) - \beta e\right).$$

To evaluate the supremum in the above equation, we take the first derivative of $\mathcal{H}\left(\frac{e-\epsilon_1}{\Delta}\right) - \beta e$ with respect to energy density e and equate it to zero, to get

$$\frac{\partial \mathcal{H}\left(\frac{e-\epsilon_1}{\Delta}\right)}{\partial e}\bigg|_{e=e^*} - \beta = 0.$$

Recall that $\frac{\partial \mathcal{H}(p)}{\partial p} = \ln(1/p - 1)$ to obtain the stationary point

$$e^* = \epsilon_1 + \Delta \frac{\exp(-\beta\Delta)}{1 + \exp(-\beta\Delta)}.$$

Since $\mathcal{H}(p)$ is a concave function of p, it follows that e^* corresponds to the unique maxima. Substituting this back into the expression for the free energy, we get

$$f(\beta) = -\frac{1}{\beta}\left(\mathcal{H}\left(\frac{e^* - \epsilon_1}{\Delta}\right) - \beta e^*\right) = \epsilon_1 - \frac{1}{\beta}\log(1 + \exp(-\beta\Delta)).$$

This expression is identical to the free energy for a single particle as expected, since summation of energy functions amounts to no interaction system. The free energy of non-interacting N particle system, is the aggregate free energy of N independent single particle systems.